Maharashtra State Board of Technical Education, Mumbai

LABORATORY PLAN (LP)

Academic Year: 2025-26

Date: 13/12/2025

K-2

Institute Name & Code: K. K. Wagh Polytechnic, Nashik-3 (0078) Class: FYCH

Program and Code: Chemical Engineering (CH)

Course Index: CO404

Course Name: Fundamentals of Chemical Engineering & Material Course Code &. Abbr:312341(FCEM)

Total Hrs: 45 Semester: 2nd Scheme: K Name of Faculty: Dr.P.S.Bhandari

INDUSTRY EXPECTED OUTCOME

Chemical engineering student will be conversant with terminologies used and duties of chemical Engineer.

• COURSE LEVEL LEARNING OUTCOMES (COS)

CO204.1: Enumerate the role & responsibility of chemical engineer. .

CO204.2: Use different safety norms, symbols for performing various safe operations/processes in given Chemical industry Apply law of conservation of mass and energy to the flowing fluids.

CO204.3: Prepare the solution of given molarity/molality/normality for chemical reaction

CO204.4: Select the relevant unit operations and unit processes for given chemical industry.

CO204.5: Select suitable material of construction for relevant chemical process.

• Teaching and Examination Scheme:

Ī						Learnii	ng Sch	neme		Assessment Scheme												
	Code	<u>e</u>		gory	Acti	Actual Contact Hrs/Week					n(h)	Theory				Based on LL & TSL Practical				Based on SL		
	Course (Course Title		Course Categ	CL	TL	TT	SLH	NLH	Credits	er Duration(h)	FA- TH	SA- TH	To	otal	FA	-PR	SA-	PR	SL	A	Total Marks
		J		္သ			I				Paper	Max	Max	Max	Min	Max	Min	Max	Min	Max	min	Tc
	31234	Fundamental chemical engineering &materials	FCEM	DSC	3	-	4	1	8	4	03	30	70	10 0	40	50	20	25@	10	25	10	200

Abbreviations: CL- Class Room Learning , TL- Tutorial Learning, LL-Laboratory Learning, SLH-Self Learning Hours, NLH-Notional Learning Hours, FA - Formative Assessment, SA -Summative assessment, IKS - Indian Knowledge System, SLA - Self Learning Assessment

Legends: @ Internal Assessment, # External Assessment, *# On Line Examination, @\$ Internal Online Examination

• Laboratory Learning Outcome (LLO)

LLO No.	Title of LLO						
LLO 1.1	1.1 Enlist different types chemicals used in laboratory.						
	1.2 Describe typical technical specifications and grades of chemicals used.						
LLO 2.1	2.1 Identify hazard by referring given GHS symbol.						
LLO 3.1	3.1 Identify given class of fire.						
	3.2 Select appropriate fire extinguisher for given situation						
LLO 4.1	4.1 Select appropriate Personal Protective equipment						
LLO 5.1	5.1 Describe the concept of molarity to estimate the quantities of solute required.						
	5.2 Prepare the solution of required molarity by following standard procedure.						
	5.3 Calculate the molarity of given solution by following standardization procedure.						

LLO No.	Title of LLO
LLO 6.1	6.1 Explain the concept of normality to estimate the quantities of solute required.
	6.2 Prepare solution of given normality by following standard procedure.
	6.3 Calculate normality of solution by following standardization procedure.
LLO 7.1	7.1 Explain the concept of molality to estimate the quantities of solute and solvent required.
	7.2 Prepare solution of given molality by following standard procedure.
	7.3 Calculate molality of solution by following standardization procedure.
LLO 8.1	8.1 Explain the concept of pH.
	8.2 Demonstrate the process of pH measurement using pH meter.
	8.3 Calculate the pH by adjusting acidity or alkalinity of solution
LLO 9.1	9.1 Measure concentration of given salt solution.
	9.2 Use conductivity meter for measuring electrical conductivity of a solution.
LLO 10.1	10.1 Explain the concept of density/specific gravity.
	10.2 Measure density/specific gravity of given solution by using specific gravity
	bottle/hydrometer
LLO 11.1	11.1 Explain the concept of DBT and WBT.
	11.2 Describe construction and working of sling psychrometer.
	11.3 Use sling psychrometer for measuring DBT and WBT
LLO 12.1	12.1 Explain the concept of size separation by using screening.
	12.2 Use screening operation for separating given sample as per particle size range
LLO 13.1	13.1 Explain the concept of saturation solubility.
	13.2 Prepare saturated solution of given solute and determine saturation solubility
LLO 14.1	14.1 Explain the concept of leaching operation.
	14.2 Demonstrate the leaching operation.
LLO 15.1	15.1 Explain the concept of corrosion.
	15.2 Measure the rate of corrosion in different environment
LLO 16.1	16.1 Explain the effect of medium on rate of corrosion.
	16.2 Describe the concept of corrosion resistance of the material.

• COs, Practical Laboratory Learning Outcome (LLOs) and Mapping:

PR.	nt COs	Practical Laboratory	Name of Experiments/Assignment/	Plar	nned Dates	ate of	Remark/ Assessment
No	Relevant COs	Learning Outcome (LLO)	Sheet/ Job/ Project Activity	From	То	Actual Dat conduction	Date with Staff sign
		LLO 1.1	*Prepare a list of chemical	18/12/2025	25/12/2025		
1		LLO 1.2	available in the chemical lab	19/12/2025	26/12/2025		
			consisting of details like manufacture, Grade of chemical				
			[LR/AR /HPLC] % purity,				
	CO1		Specific gravity, CAS no.snd				
	5		Chemical formula.	20/12/2025	27/12/2025		
		LLO 2.1	* C	25/12/2025	01/01/2026		
2	1	LLO 2.2	* Categorize hazard for given chemicals using the GHS symbols.	26/12/2025	02/01/2026		
	C01			27/12/2025	03/01/2026		

PR.	nt COs	Practical Laboratory	Name of Experiments/Assignment/	Plan	ned Dates	Actual Date of conduction	Remark/ Assessment Date with Staff sign
No	Relevant COs	Learning Outcome (LLO)	Sheet/ Job/ Project Activity	From	То		
		LLO 3.1	* Demonstrate the use of fire	01/01/2026	08/01/2026		
3	C02	LLO32	extinguishers available in the laboratory.	02/01/2026	09/01/2026		
	ŭ		3	03/01/2026	10/01/2026		
	7	LLO 4.1	* Demonstrate the use of personal	08/01/2026	15/01/2026		
4	C02	LLO 4.2	*.Demonstrate the use of personal protective equipment's	09/01/2026	16/01/2026		
		•	processive equipment b	10/01/2026	17/01/2026		
	61	11051	* D	15/01/2026	22/01/2026		
5	C02	LLO 5.1 LLO 5.2	* Prepare and standardize a solution of given molarity	16/01/2026	23/01/2026		
		LEO 3.2		17/01/2026	24/01/2026		
	CO3	LLO 7.1 LLO 7.2	* Prepare and standardize a solution of given normality	22/01/2026	29/01/2026		
6				23/01/2026	30/01/2026		
				24/01/2026	31/01/2026		
	4	LLO 10.1 LLO 106.2	*Prepare solution of given pH.	29/01/2026	05/02/2026		
7	C04			30/01/2026	06/02/2026		
				31/01/2026	07/02/2026		
	4	LLO 11.1 LLO 11.2	* Calculate the density of various concentration to measure electrical	05/02/2026	12/02/2026		
8	C04			06/02/2026	13/02/2026		
				07/02/2026	14/02/2026		
	05	LLO 13.1 LLO 13.2	I sting psychrometer to stildy the	12/02/2026	19/02/2026		
9	2			13/02/2026	20/02/2026		
				14/02/2026	21/02/2026		
	10	110151	* Use given salt to prepare to prepare saturated solution	19/02/2026	26/02/2026		
10	CO5	LLO 15.1 LLO 15.2		20/02/2026	27/02/2026		
				21/02/2026	28/02/2026		
	w	LLO 14.1	* Calculate % recovery of dye from	26/02/2026	05/03/2026		
11	CO5	LLO 14.1 LLO 14.2	beet root or other substrate by leaching operation	27/02/2026	06/03/2026		
			U 1	28/02/2026	07/03/2026		
	ıc	110161	* Calculate the rate of corrosion in	05/03/2026	12/03/2026		
12	CO5	LLO 16.1 LLO 16.2	Acidic/Alkaline/Saline medium	06/03/2026	13/03/2026		
				07/03/2026	14/03/2026		

• ASSESSMENT METHODOLOGIES/TOOLS

A. Formative assessment (Assessment for Learning) (FA-TH)

- Continuous assessment based on process and product related performance indicators. Each practical will be assessed considering
 - o 60% weightage is to process
 - o 40% weightage to product

B. Summative Assessment (Assessment of Learning) (SA-TH)

- Continuous Assessment based on Process and Product related performance indicators. Each practical will be assessed considering
 - o 60% weightage to Process
 - o 40% weightage to Product

• Laboratory Equipment / Instruments / Tools / Software required

Sr.	Equipment Name with Broad Specifications	Relevant LLO
No.		Number
1	Respiratory and Non-respiratory Personal Protective Equipment as per IS standards	1,2
2	Public Address system	15
3	Portable Fire Extinguishers (CO2, Foam, Dry powder, clean extinguishing agent) as per IS standards.	3
4	Fire hydrant system	4

References:

• Suggested Learning Materials / Books

Sr. No.	Author	Title of Book	Publication
1	Dr. K. U. Mistry	Fundamentals of Industrial Safety and Health	Siddhant Prakashan, Ahmedabad, Gujrat
2	Crowl, Daniel A, Louvar, Joseph F.	Chemical Process Safety	Prentice Hall, NJ, USA, 2002,ISBN 0-13-018176-5
3	Bureau of Indian Standards	IS 14489: 1998	Government of India.
4	Bureau of Indian Standards	IS 17889: 2022	Government of India.
5	Bureau of Indian Standards	IS 17893:2023	Government of India.
6	Department of Environment, Forest and wildlife.	The Manufacture, Storage, and Import of Hazardous Chemical Rules, 1989	Government of India.

• Learning Websites & Portal

Sr. No	Link / Portal	Description
1	https://onlinecourses.swayam2.ac.in/nou23_ge81/preview	Fire prevention and protection
2	https://onlinecourses.nptel.ac.in/noc20_mg43/preview	Functioning in safer way
3	https://archive.nptel.ac.in/courses/103/106/103106071/	Fire and Explosion
4	https://onlinecourses.nptel.ac.in/noc22_ch44/preview	General chemical safety measures
5	https://safetyculture.com/topics/ppe-safety/	Personal Protective Equipment(PPE)

Dr. P.S.Bhandari (Name & signature of staff) Dr. P. S. Bhandari (Name & signature of HOD)